#### Protein Structure 101

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## Proteins play key roles in a living system

- Three (out of many many) examples of protein functions
  - Catalysis: Almost all chemical reactions in a living cell are catalyzed by protein enzymes.
  - Transport: Some proteins transports various substances, such as oxygen, ions, and so on.
  - Information transfer: For example, hormones.



Alcohol dehydrogenase oxidizes alcohols to aldehydes or ketones

Myoglobin stores oxygen





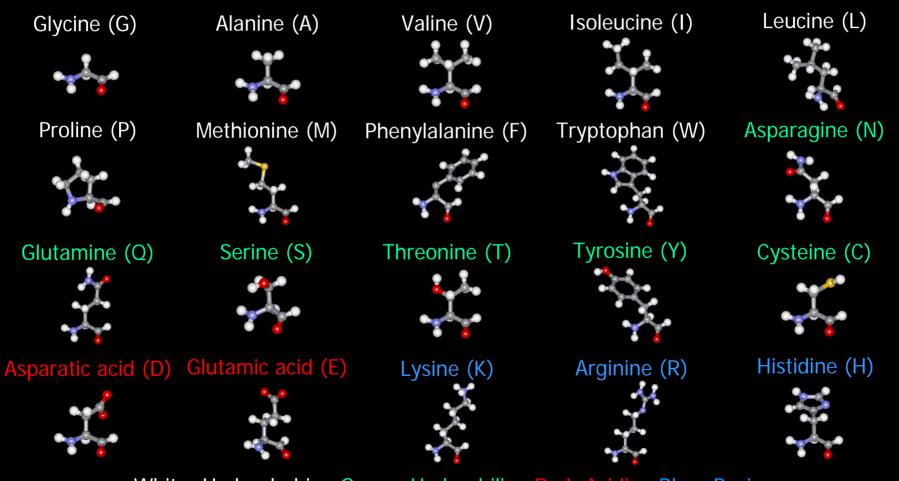
Insulin controls the amount of sugar in the blood

### Amino acid: Basic unit of protein

R I NH<sub>3</sub>+-C-COO Amino group H Different side chains, R, determin the properties of 20 amino acids.

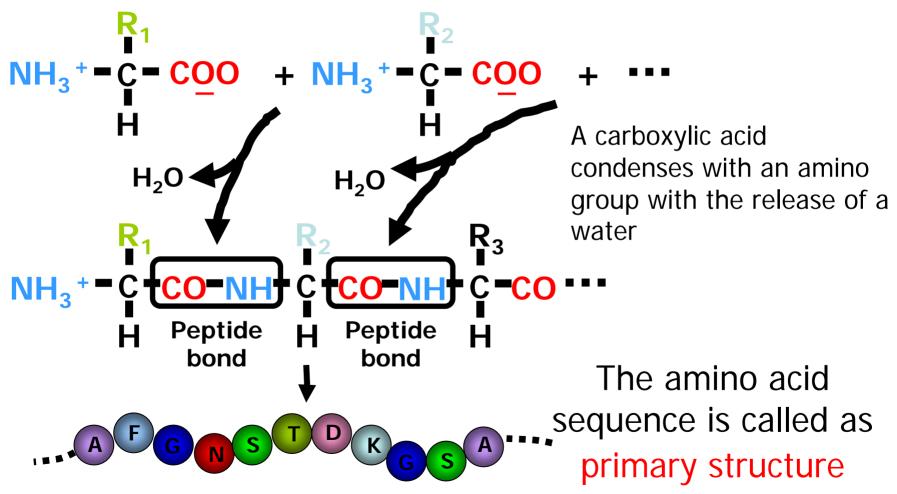
An amino acid

### 20 Amino acids



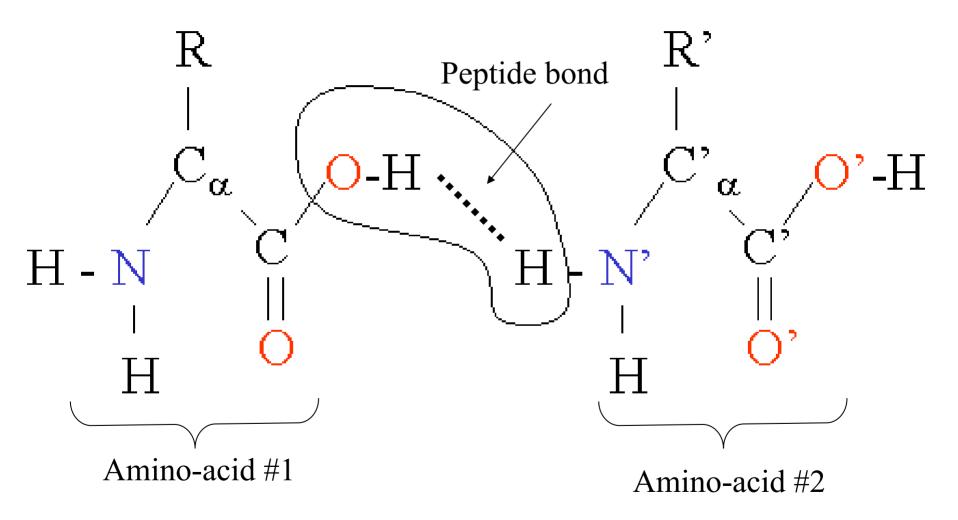
White: Hydrophobic, Green: Hydrophilic, Red: Acidic, Blue: Basic

### Proteins are linear polymers of amino acids



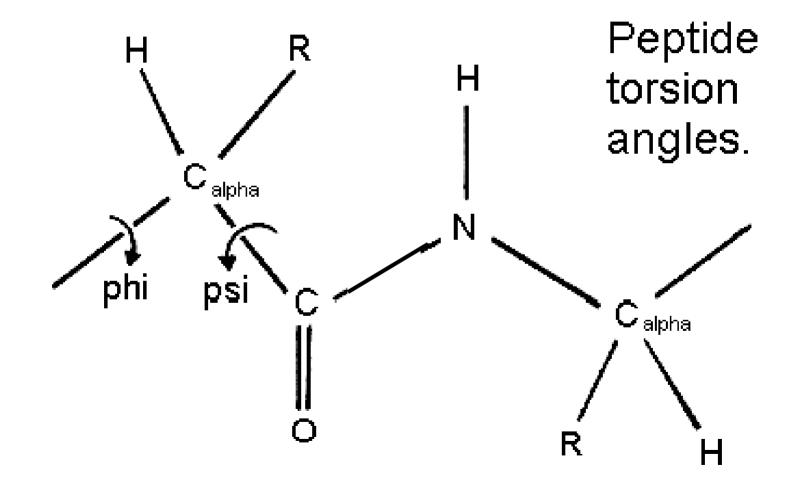
#### Protein Structure in 3 steps.

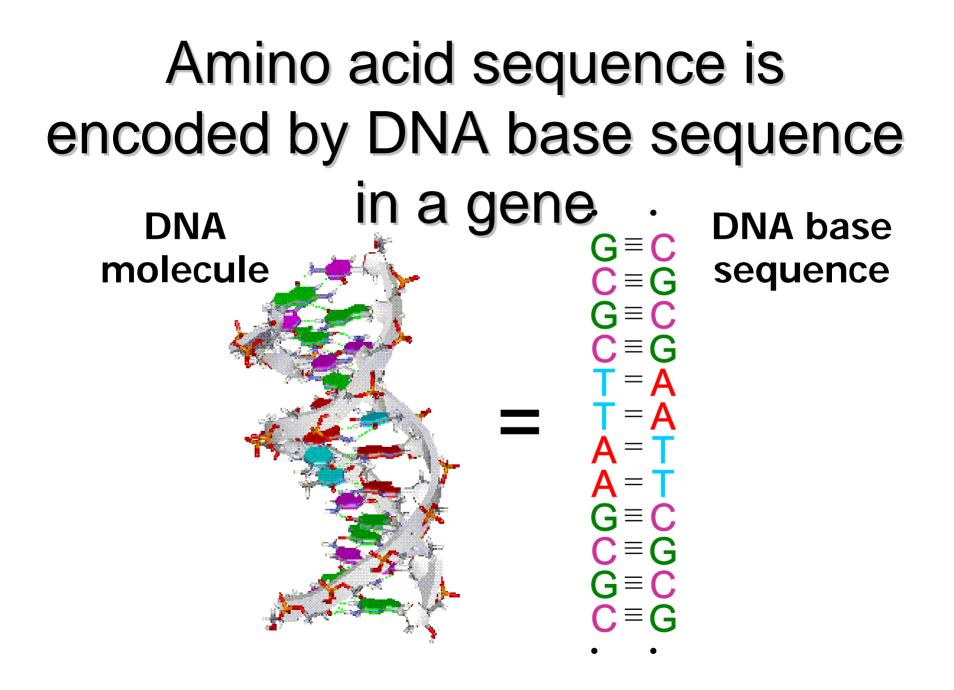
Step 1. Two amino-acids together (di-peptide)



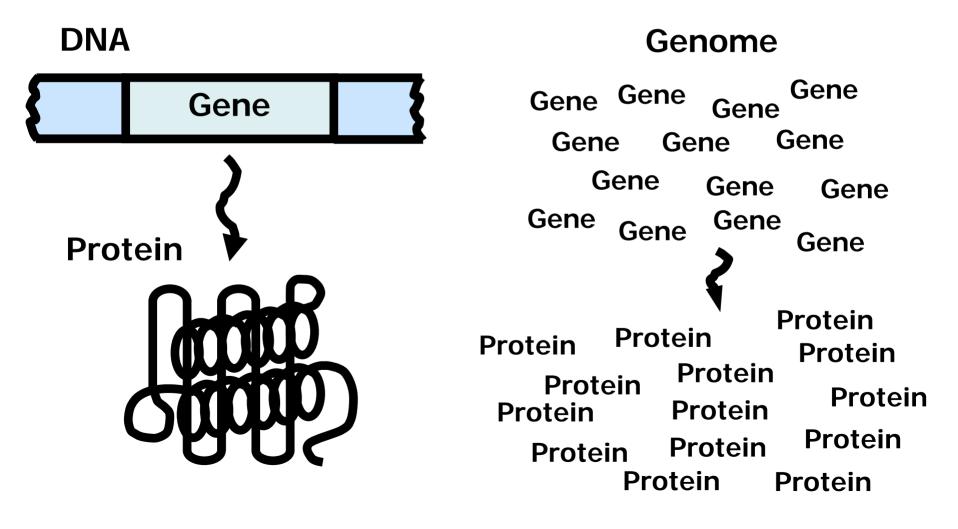
Protein Structure in 3 steps.

Step 2: Most flexible degrees of freedom:

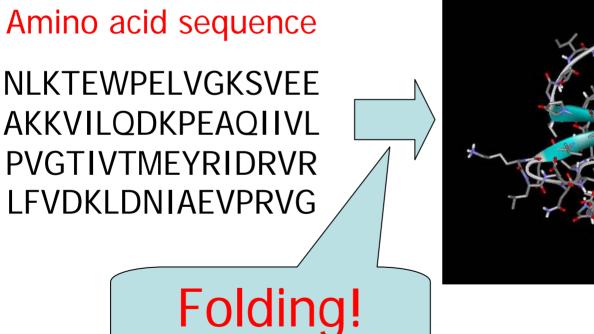


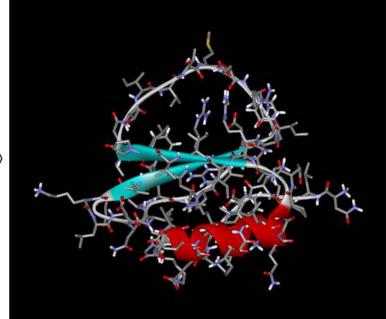


# Gene is protein's blueprint, genome is life's blueprint

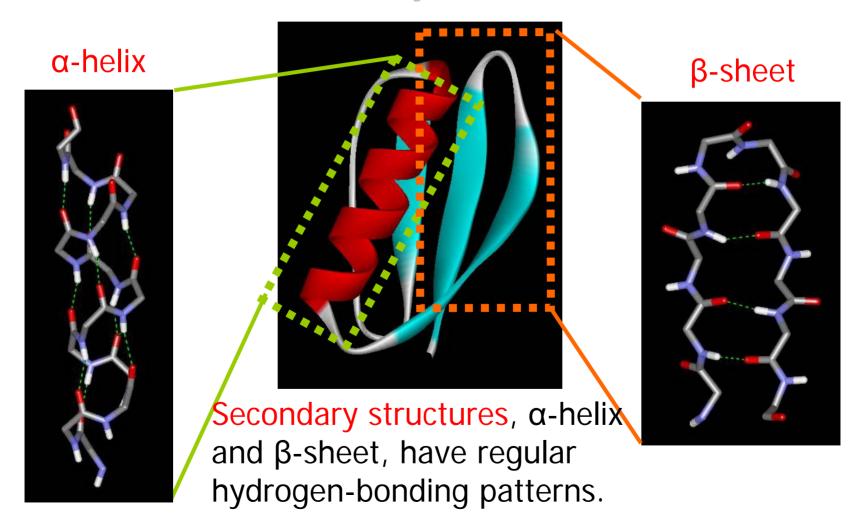


# Each Protein has a unique structure





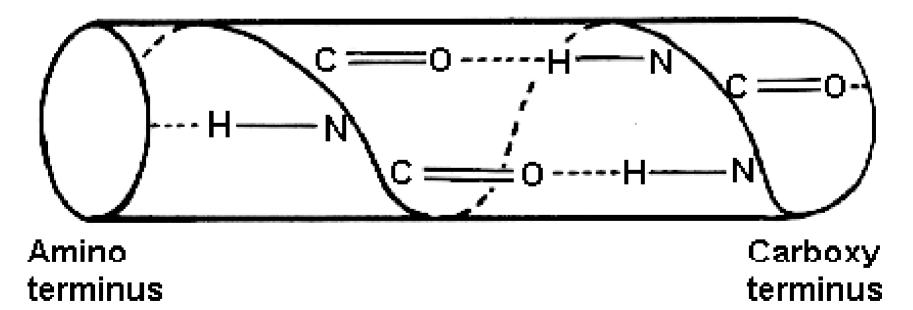
### Basic structural units of proteins: Secondary structure



#### Protein Structure in 3 steps.

Sometimes, polypeptide chain forms helical structure:

Toilet roll representation of the main chain hydrogen bonding in an alpha-helix.



## Hydrogen Bonding

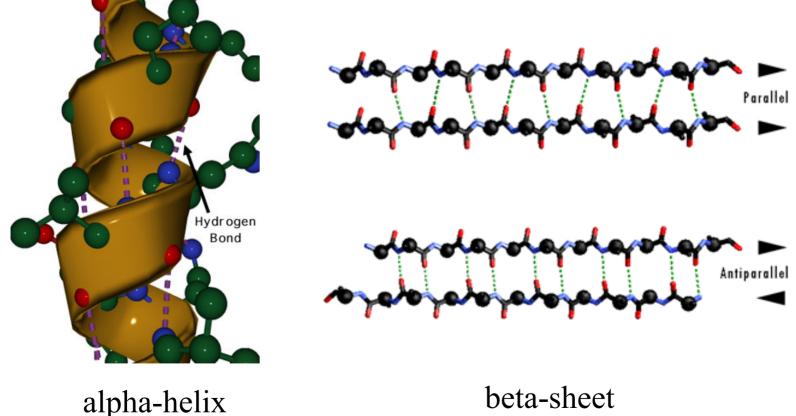
- Involves three atoms:
  - Donor electronegative atom (D) (Nitrogen or Oxygen in proteins)
  - Hydrogen bound to donor (H)
  - Acceptor electronegative atom (A) in close proximity

#### $\mathbf{D} - \mathbf{H} = \mathbf{A}$

#### **D-H** Interaction

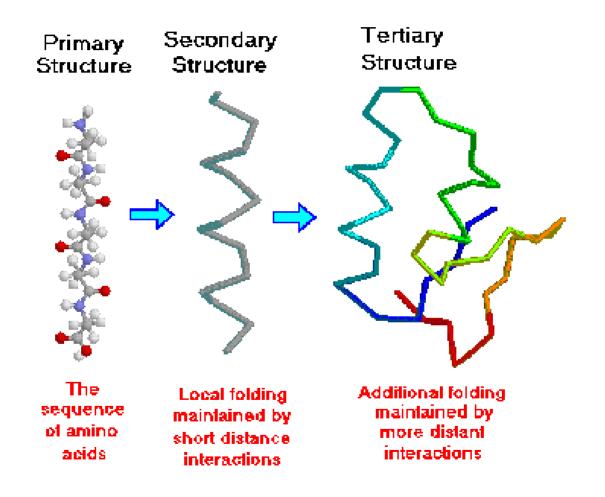
- Polarization due to electron withdrawal from the hydrogen to D giving D partial negative charge and the H a partial positive charge
- Proximity of the Acceptor A causes further charge separation  $\delta \delta + \delta \delta$ 
  - $\mathbf{D} \mathbf{H} = \mathbf{A}$
- Result:
  - Closer approach of A to H
  - Higher interaction energy than a simple van der Waals interaction

#### Hydrogen Bonding And Secondary Structure

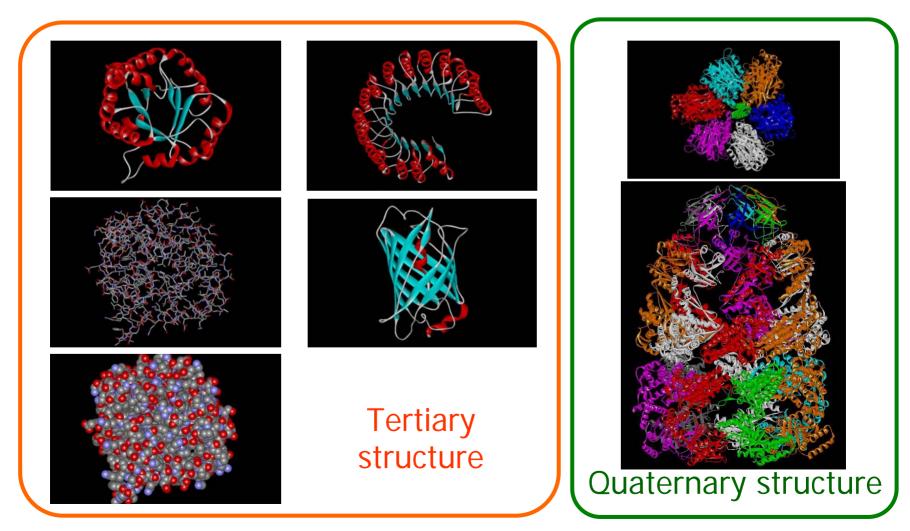


beta-sheet

#### **Protein Structure**



# Three-dimensional structure of proteins



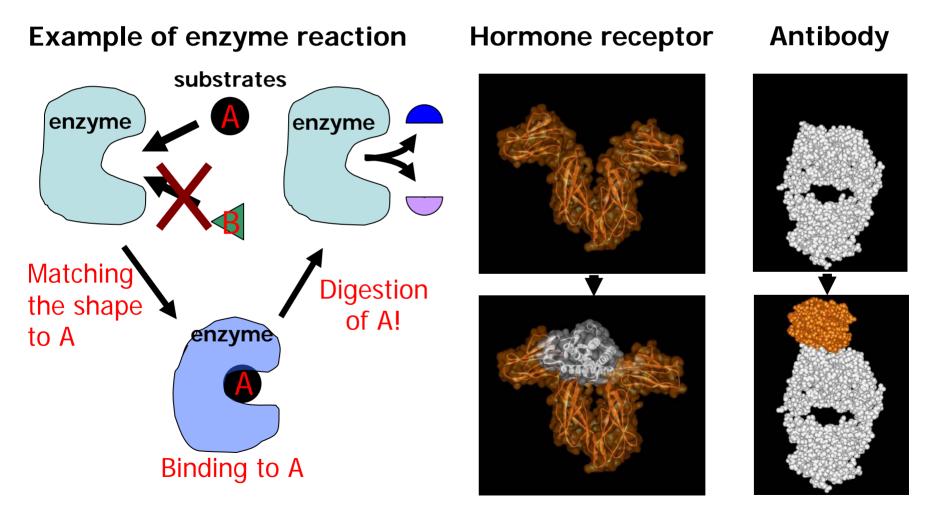
# Hierarchical nature of protein structure

Primary structure (Amino acid sequence)  $\downarrow$ Secondary structure ( $\alpha$ -helix,  $\beta$ -sheet)

Tertiary structure (Three-dimensional structure formed by assembly of secondary structures )

Quaternary structure (Structure formed by more than one polypeptide chains )

# Close relationship between protein structure and its function



Protein structure prediction has remained elusive over half a century

"Can we predict a protein structure from its amino acid sequence?"

Still virtually impossible at atomic level accuracy (but there are some notable exceptions). Possible in some cases if a rougher structure is acceptable.

#### So where do we get the high quality protein structures to work with?

- THE PDB (Protein Data Bank. ~30,000 structurs)
- <u>PDB</u>

### Summary

- Proteins are key players in our living systems.
- Proteins are polymers consisting of 20 kinds of amino acids.
- Each protein folds into a unique three-dimensional structure defined by its amino acid sequence.
- Protein structure has a hierarchical nature.
- Protein structure is closely related to its function.